

THE STRUCTURE OF PEDICULIDINE

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From the epigeal part of *Pedicularis olgae* collected on June 18, 1968 (region of the village of Sagyrdasht, TadzhSSR), 0.59% of total bases have been isolated by chloroform extraction. The ethereal fraction of the combined alkaloids yielded the picrate of a base with mp 211–212°C (ethanol), which was chromatographed on alumina. A base was isolated with the composition C₁₀H₉NO, mp 74–75°C, M⁺ 159 (mass spectrometrically), R_f 0.51. The IR spectrum of the base had the three maxima at λ_{max} 268, 273 and 293 nm (log ε 3.97, 39.6, and 3.36) that are characteristic for alkaloids of the pyridine type [1, 2]. The

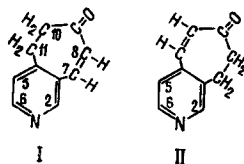
IR spectrum of the alkaloid had absorption bands at (cm⁻¹) 1695–1640 (—C=O—CH=CH—), 1620, 1590, 1595 (pyridine ring), and 810, 855 (1,3,4-trisubstituted benzene ring) [3]. The base proved to be new, and we have called it pediculidene (I).

In the NMR spectrum of (I) (δ scale) (Fig. 1), a one-proton singlet appears clearly in the weak-field region at 8.51 ppm, corresponding to the hydrogen atom at C-2, and there are two one-proton doublets at 8.41 and 7.15 ppm, J=5.0 Hz, corresponding to two atoms of hydrogen at C-5 and C-6 of a pyridine ring. The absence of other signals in this region shows that positions 3 and 4 of the pyridine ring of (I) are substituted.

Two olefinic protons of C-7 and C-8 are observed in the form of two symmetrical one-proton doublets with centers at 7.12 and 6.34 ppm, J=12.2 Hz. The magnitude of the spin-spin coupling constant of the olefinic protons shows that the latter are in the cis position in a seven-membered ring [4–7]. Signals in the 2.45–3.15 ppm region correspond to two methylene protons at C-10 and C-11.

The mass spectrum of the base has peaks of the ions M⁺ (159) and 158, 132, 131, 130, 118, 117, 104, 103, 102, 91, 89, 77 m/e. This manner of fragmentation is characteristic for alkaloids of the pyridine type [8].

On the basis of the facts presented, two structural formulas may be proposed for pediculidene: (I) or (II). Biogenetic considerations [9] permit the choice of formula (I) as the more likely.



EXPERIMENTAL

The NMR spectra were taken on a JNM-4-H-100/100 MHz instrument (in CDCl₃), and the mass spectra on an MKh-1303 instrument with a glass inlet system at an ionizing voltage of 40 eV.

Isolation of Pediculidene (I). The comminuted epigeal part of *P. olgae* (30 kg) was moistened with 10% ammonia solution (1:1) and charged into an extractor. After it had stood for 2 hours, it was covered with chloroform and extraction was carried out overnight. A total of 12 overflows took place. The con-

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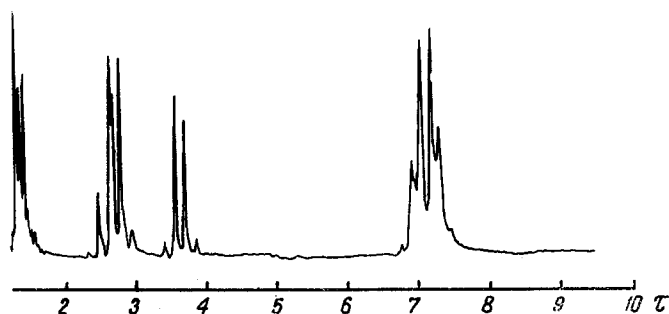


Fig. 1. NMR spectrum of pediculidine.

centrated chloroform extract was treated with 10% sulfuric acid. The combined acid solution was washed with ether and filtered and, with cooling, it was made alkaline with 25% ammonia solution. The alkaloids were extracted first with ether and then with chloroform. Distillation of the ether yielded 125.3 g of combined ether alkaloids and the chloroform solution gave 51.7 g of combined chloroform alkaloids. The total weight of the alkaloids was 177 g (0.59% of the weight of the dry plant).

The ether alkaloids (15 g) were treated with petroleum ether. Into this passed 6 g of the ether alkaloids, which was dissolved in 40 ml of acetone and treated with a saturated ethanolic solution of picric acid.

This gave 1.8 g of the picrate of an alkaloid with mp 204–208°C; after recrystallization from ethanol, mp 211–212°C. Yield 1.25 g. The picrate (1.25 g) was converted into the base by chromatography on a column of alumina (50 g; activity grade II). From the chloroform eluate 0.45 g of crystals with mp 70–74°C was isolated; after sublimation, mp 74–75°C; R_f 0.51 on TLC in silica gel in the methanol–chloroform–butyl acetate (1:2:1) system.

SUMMARY

A new alkaloid pediculidine, $C_{10}H_9NO$, mp 74–75°C, has been isolated from the ethereal fraction of the combined alkaloids of *P. olgae*.

Structure (I) has been proposed for the new alkaloid on the basis of its UV, IR, NMR, and mass spectra.

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